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## INDIAN SCHOOL MUSCAT SECOND PERIODIC ASSESSMENT

### CHEMISTRY

CLASS: 12

Sub.Code: 043

Time Allotted: 30 mts.

03 .05.2021

Max .Marks: 20

#### GENERAL INSTRUCTIONS:

- a. All questions are compulsory.
- b. Mark for each question is indicated against it

Following questions are multiple choice type carrying 1 mark each:

1. Which of the following reagents convert propene to Propan-1-ol? 1  
 a)  $\text{H}_2\text{O}$ ,  $\text{H}_2\text{SO}_4$                       b) Aqueous KOH  
 c)  $\text{MgSO}_4$ ,  $\text{NaBH}_4/\text{H}_2\text{O}$               d)  $\text{B}_2\text{H}_6$ ,  $\text{H}_2\text{O}_2/\text{OH}^-$
2. An organic compound X of molecular formula  $\text{C}_4\text{H}_{10}\text{O}$  undergoes oxidation with acidified  $\text{KMnO}_4$  to give a compound Y of molecular formula  $\text{C}_4\text{H}_8\text{O}_2$ . X could be 1  
 a)  $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$               b)  $\text{CH}_3\text{CH}_2\text{CH}(\text{OH})\text{CH}_3$   
 c)  $\text{CH}_3\text{CH}=\text{CHCH}_2\text{OH}$               d)  $(\text{CH}_3)_3\text{COH}$
3. Picric acid is: 1  
 a) Trinitrophenol                      b) Trinitrotoluene  
 c) Trinitrobenzene                      d) Tribromobenzene

**In the following questions, a statement of assertion followed by a statement of reason is given. Choose the correct answer out of the following choices.**

- A. Both Assertion (A) and Reason (R) are correct statements, and Reason (R) is the correct explanation of the Assertion (A).
  - B. Both Assertion (A) and Reason (R) are correct statements, but Reason (R) is **not** the correct explanation of the Assertion (A).
  - C. Assertion (A) is correct, but Reason (R) is incorrect statement.
  - D. Assertion (A) is incorrect, but Reason (R) is correct statement
4. Assertion: The  $\text{pK}_a$  of ethanol is lower than that of phenol 1  
 Reason: Phenoxide ion is resonance stabilised whereas ethoxide ion is not.

5. Assertion: Ethanol gives the same ester as product when reacted with ethanoic acid, ethanoyl chloride and ethanoic anhydride. 1  
Reason: Ethanoic acid, ethanoyl chloride and ethanoic anhydride are all acetylating reagents.
6. Assertion: Bond angle in ethers is greater than the tetrahedral angle. 1  
Reason: Steric repulsion between the alkyl groups is greater than the lone pair lone pair repulsions.

**Fill in the blanks**

7. IUPAC name of  $\text{CH}_3\text{C}(\text{Br})=\text{CHCH}_2\text{OH}$  is \_\_\_\_\_ 1
8.  $\text{C}_6\text{H}_5-\text{OH} \xrightarrow[\text{(ii) H}^+]{\text{(i) CHCl}_3 + \text{aq. NaOH}}$  1

**Answer the following**

9. Give a chemical test to distinguish ethanol and phenol. 1
10. Arrange the following in the increasing order of acidic nature. 1  
2,4,6 – trinitrophenol, n-butyl alcohol, 3,5 – dinitrophenol, 4-methylphenol, tert-butyl alcohol, 2,4-dinitrophenol, ethanol.
11. Reaction of alcohol with acid chloride is carried out in presence of a pyridine. Give reason 1
12. Give the structures of final products expected from the following reactions. 2
  - a) Heating of benzyl phenylether with HI
  - b) Phenol is treated with  $\text{Na}_2\text{Cr}_2\text{O}_7$  and con  $\text{H}_2\text{SO}_4$
13. Illustrate the following with suitable chemical equations 2
  - a) Kolbe's reaction
  - b) Williamson's synthesis
14. Convert 2
  - a) Benzene to phenol
  - b) Ethanal to sec-butyl alcohol
15. Explain the mechanism of acid catalyzed hydration of ethene. 3